

that are enhanced by heme–His(F8) interactions. Moreover, because HbO₂, HbCO, and metHb lack the modes at $\sim 250\text{ cm}^{-1}$ observed in the corresponding Mb species, their enhancement must be associated with differences in heme–histidine conformation between Mb and Hb. Similar arguments hold for the native and partially unfolded (low pH) forms of MbCO. We tentatively suggest that the intensity of the $\sim 250\text{-cm}^{-1}$ mode is a probe of

the eclipse angle between the histidine plane and the N_p–N_p axis.

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Registry No. Fe, 7439-89-6; His, 71-00-1; heme, 14875-96-8.

Communications to the Editor

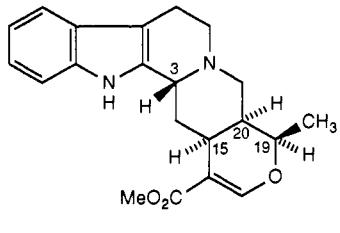
Enantioselective Synthesis of (+)-3-Isorauniticine via a Catalytic Tandem “Palladium-Ene”/Carbonylation Reaction

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The heteroyohimbine alkaloid 3-isorauniticine, isolated from *Corynanthe mayumbensis*, has been shown to possess constitution and relative configuration **1**.¹ We present here the first total synthesis of (+)-**1**,² thereby assigning its absolute configuration.³



The cornerstone of our strategy is an intramolecular Pd-catalyzed allylation/carbonylation process, recently employed for a synthesis of (\pm)-pentalenolactone **E**.⁴ For the preparation of **1** we envisaged control, in this key step, of the configuration of developing centers C(15) and C(20) by means of a preexisting

(1) Melchio, J.; Bouquet, A.; Pa  s, M.; Goutarel, R. *Tetrahedron Lett.* 1977, 315–316.

(2) For syntheses of other, optically pure or racemic, heteroyohimbines, see: (a) Uskokovic, M. R.; Lewis, R. L.; Partridge, J. J.; Despreaux, C. W.; Pruess, D. L. *J. Am. Chem. Soc.* 1979, 101, 6742–6744. (b) Massiot, G.; Mulamba, T. *J. Chem. Soc., Chem. Commun.* 1984, 715–716. (c) Hatakeyama, S.; Saijo, K.; Takano, S. *Tetrahedron Lett.* 1985, 26, 865–868. (d) Takano, S.; Satoh, S.; Ogasawara, K. *J. Chem. Soc., Chem. Commun.* 1988, 59–60. (e) Hirai, Y.; Terada, T.; Yamazaki, T. *J. Am. Chem. Soc.* 1988, 110, 958–960. (f) Baggolini, E. G.; Pizzolato, G.; Uskokovic, M. R. *Tetrahedron* 1988, 44, 3203–3208. (g) H  scher, P.; Kn  lker, H.-J.; Winterfeldt, E. *Tetrahedron Lett.* 1990, 31, 2705–2706. (h) Hanessian, S.; Faucher, A.-M. *J. Org. Chem.* 1991, 56, 2947–2949. (i) Winterfeldt, E.; Radunz, H.; Korth, T. *Chem. Ber.* 1968, 101, 3172–3184. (j) Winterfeldt, E.; Gaskell, A. J.; Korth, T.; Radunz, H.-E.; Walkowiak, M. *Chem. Ber.* 1969, 102, 3558–3572. (k) van Tamelen, E. E.; Placeway, C.; Schiemann, G. P.; Wright, I. G. *J. Am. Chem. Soc.* 1969, 91, 7359–7371. (l) Gutzwiller, J.; Pizzolato, G.; Uskokovic, M. R. *Helv. Chim. Acta* 1981, 64, 1663–1671; *J. Am. Chem. Soc.* 1971, 93, 5907–5908. (m) Wenkert, E.; Chang, C.-J.; Chawla, H. P. S.; Cochran, D. W.; Hagaman, E. W.; King, J. C.; Orito, K. *J. Am. Chem. Soc.* 1976, 98, 3645–3655. (n) Loukasmaa, M.; Jokela, R. *Tetrahedron Lett.* 1986, 27, 2043–2044. (o) Naito, T.; Kojima, N.; Miyata, O.; Ninomiya, I.; Inoue, M.; Doi, M. *J. Chem. Soc., Perkin Trans. I* 1990, 1271–1280.

(3) All heteroyohimbines were postulated to have the C(15) configuration as depicted in **1**. For the partial conversion (4%) of hirsuteine to 3-isorauniticine, see: Sakai, S.-I.; Shinma, N. *Chem. Pharm. Bull.* 1978, 26, 2596–2598.

(4) Oppolzer, W.; Xu, J.-Z.; Stone, C. *Helv. Chim. Acta* 1991, 74, 465–468.

center C(3).⁵ To set up center C(3) we took advantage of a convenient multigram approach to enantiomerically pure α -amino acids (Scheme I).⁶

Thus, C-alkylation of commercially available chiral glycinate equivalent **2**^{6,7} with allyl iodide/LiOH under phase-transfer conditions^{6b} followed by acidic removal of the N-protecting group and N-acylation with mesitylenesulfonyl chloride provided crystalline sulfonamide **3**⁸ (69% from **2**, mp 151–152 °C). N-Alkylation of **3** with (*Z*)-1-bromo-4-[(methoxycarbonyl)oxy]-2-butene⁹ furnished dienylcarbonate **4** (96% yield).⁸ Proceeding to the key reaction, carbonate **4** was subjected to Pd(0)-catalyzed cyclization/carbonylation in acetic acid, giving a 67:22:11 mixture of **5** and two stereoisomers.¹⁰ Flash chromatography (FC) afforded the crystalline major isomer **5**⁸ (mp 190–191 °C) in 45–53% yield.

Catalytic hydrogenation of **5** from the less hindered exo face and subsequent Baeyer–Villiger oxidation yielded lactone **6**⁸ (mp 141–143 °C, 86% from **5**). We now needed to deprotect first the amino group and then the carboxyl group without affecting the lactone moiety. “Transesterification” of acyl sultam **6** with lithium *p*-nitrobenzyl oxide and FC gave the *p*-nitrobenzyl ester **7**⁸ (55%, mp 118–119 °C). Starting from **7**, successive cleavage of the sulfonamide (HF/pyridine¹¹), N-alkylation with tryptophyl bromide, and hydrogenolysis furnished carboxylic acid **8**⁸ (62% from **7**), which was subjected to a PhPOCl₂-mediated Rapoport cyclization,¹² giving pentacyclic lactone **9**⁸ (46%, mp 268–271 °C dec) as a single stereoisomer. The transformation **8** → **9** apparently involves decarbonylation of **8** with loss of the C(3) configuration, which is reestablished in the subsequent Pictet–Spengler step.

Finally, formylation of lactone **9** followed by Korte “rearrangement”^{2a,c,g,j,13} provided (+)-3-isorauniticine (53% from **9**, hydrochloride: mp 258–260 °C dec, $[\alpha]_D = +37.4^\circ$ ($c = 0.77$, MeOH, $T = 19.5$ °C); lit.¹ mp 277 °C, lit.¹ $[\alpha]_D = +25^\circ$ ($c = 1$, MeOH)). The ¹H NMR, ¹³C NMR, IR, and CD spectra of synthetic **1** are identical with those of the naturally occurring alkaloid.¹⁴

(5) For the topological influence of preexisting centers on “palladium-ene” cyclizations, see: Oppolzer, W.; Keller, T. H.; Kuo, D. L.; Pachinger, W. *Tetrahedron Lett.* 1990, 31, 1265–1268.

(6) (a) Oppolzer, W.; Moretti, R.; Thomi, S. *Tetrahedron Lett.* 1989, 30, 6009–6010. (b) Oppolzer, W.; Zhou, C.; Moretti, R., publication in preparation.

(7) Oxford Chirality, PO Box 412, Oxford OX1 3QW, England.

(8) All new compounds were characterized by IR, ¹H NMR, ¹³C NMR, and mass spectra.

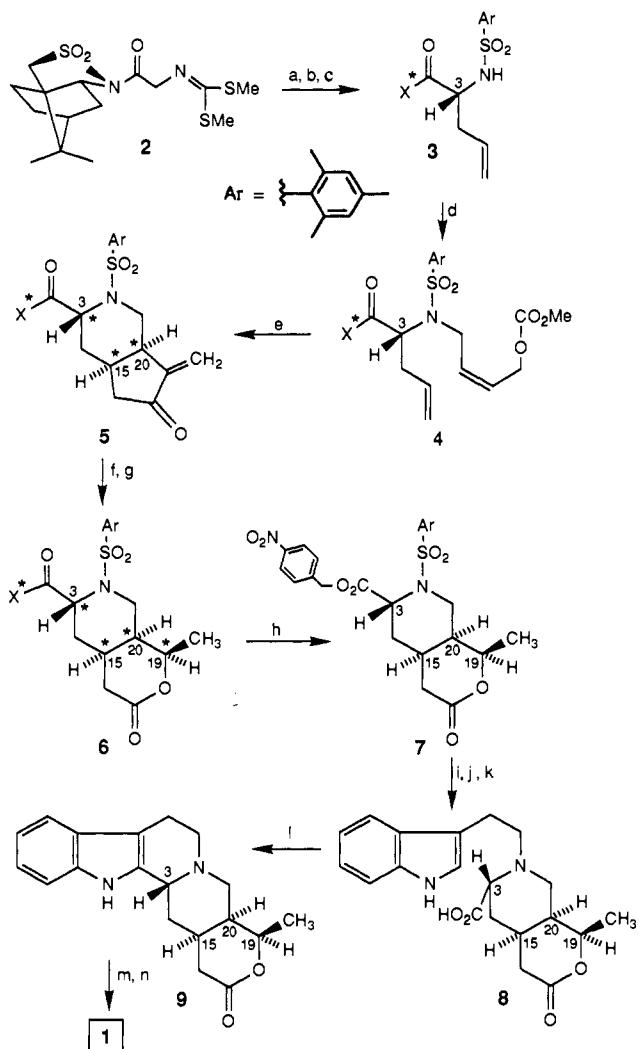
(9) Obtained by successive treatment of (*Z*)-2-butene-1,4-diol with methyl chloroformate/pyridine and CBr₄/PPh₃ (59%).

(10) For the formation of annulated methylene cyclopentanones by “palladium-ene” cyclizations, see: Yamamoto, K.; Terakado, M.; Murai, K.; Miyazawa, M.; Tsuji, J.; Takahashi, K.; Mikami, K. *Chem. Lett.* 1989, 955–958. Oppolzer, W.; Keller, T. H.; Bedoya-Zurita, M.; Stone, C. *Tetrahedron Lett.* 1989, 30, 5883–5886.

(11) Yajima, H.; Takeyama, M.; Kanaki, J.; Nishimura, O.; Fujino, M. *Chem. Pharm. Bull.* 1978, 26, 3752–3757.

(12) Johansen, J. E.; Christie, B. D.; Rapoport, H. *J. Org. Chem.* 1981, 46, 4914–4920.

(13) Korte, F.; B  chel, K. H. *Angew. Chem.* 1959, 71, 709–722.

Scheme I^a

^a (a) Allyl iodide (1.2 equiv), Bu_4NHSO_4 (1.2 equiv), LiOH (50 equiv), $\text{CH}_2\text{Cl}_2/\text{H}_2\text{O}$ (19:1), ultrasound, -7°C (bath), 6 min. (b) HCl (0.5 N), $\text{THF}/\text{H}_2\text{O}$ (1:1), room temperature, 4 h. (c) Mesitylenesulfonyl chloride (1.5 equiv), pyridine (3.5 equiv), CH_2Cl_2 , reflux, 24 h. (d) (Z)-1-Bromo-4-[(methoxycarbonyl)oxy]-2-butene (1.2 equiv), NaH (1 equiv), DMF, 0°C , 12 h. (e) $\text{Pd}(\text{dba})_2$ (0.1 equiv), PBu_3 (0.3 equiv), CO (1 atm), AcOH, 80°C , 3 h. (f) Pd/C (0.1 equiv), H_2 (1 atm), EtOAc, room temperature, 18 h. (g) MCPBA (80%, 1.5 equiv), NaHCO_3 (10 equiv), CH_2Cl_2 , room temperature, 18 h. (h) *p*-Nitrobenzyl alcohol (1.3 equiv), *n*-BuLi (1 equiv), THF/hexane (25:1), -30°C , 0.5 h, then addition of 6, $-30^\circ\text{C} \rightarrow -10^\circ\text{C}$, 6 h. (i) Pyridine/70% HF (excess), anisole (2 equiv), room temperature, 8 h. (j) Tryptophyl bromide (1.2 equiv), NaHCO_3 (10 equiv), MeCN, 80°C , 6 h; then addition of further tryptophyl bromide (0.4 equiv), 80°C , 5 h. (k) Pd/C (0.05 equiv), H_2 (1 atm), EtOH, room temperature, 0.5 h. (l) PhPOCl_2 (excess), 105°C , 4 min, then addition of 1 N aqueous HCl, 70°C , 10 min. (m) NaHMDS (10 equiv), THF, -78°C , 2 h, then addition of methyl formate (40 equiv), -78°C , 1 h, then $\rightarrow 0^\circ\text{C}$, 4 h. (n) Saturated HCl/MeOH, CH_2Cl_2 (1:9), 120°C (sealed tube), 24 h, then *p*-TsOH (5 equiv), CH_2Cl_2 , reflux, 15 h.

In summary, (+)-3-isouranidine has been synthesized via a sequence of 14 steps, which highlights once more the preparative utility of sultam-directed asymmetric alkylations^{6,15} and of transition-metal-catalyzed carbometalation/carbonylation reactions.^{4,5,10,16}

(14) Lounasmaa, M.; Kan, S.-K. *Tetrahedron* 1980, 36, 1607–1611. Morales-Rios, M. S.; Espineira, J.; Joseph-Nathan, P. *Magn. Reson. Chem.* 1987, 25, 377–395.

(15) Review: Oppolzer, W. *Pure Appl. Chem.* 1990, 62, 1241–1250.

(16) Reviews: Oppolzer, W. *Pure Appl. Chem.* 1990, 62, 1941–1948. Oppolzer, W. *Angew. Chem. Int. Ed. Engl.* 1989, 101, 39–53; *Angew. Chem., Int. Ed. Engl.* 1989, 28, 38–52.

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Supplementary Material Available: Reaction scheme, preparations, and analysis data, including mp, IR, ^{13}C NMR, and MS (9 pages). Ordering information is given on any current masthead page.

Silicon-Mediated Reductive Coupling of Aldehydes and Allylic Alcohols. A Stereoselective Synthesis of Tunicaminylluracil

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We report the development of a method for carbon–carbon bond formation between the olefinic terminus of an allylic alcohol and the carbonyl carbon of an aldehyde.¹ This coupling reaction forms the basis of a highly convergent synthesis of tunicaminylluracil (**1**),² the undecose core of the tunicamycin antibiotics (tunicamycin C, shown below, is exemplary),³ from simple carbohydrate-derived precursors.

(1) Examples of intramolecular reductive coupling reactions of ketones or aldehydes with alkenes or alkynes: (a) Stork, G.; Malhotra, S.; Thompson, H.; Uchibayashi, M. *J. Am. Chem. Soc.* 1965, 87, 1148. (b) Pradhan, S. K.; Radhakrishnan, T. V.; Subramanian, R. *J. Org. Chem.* 1976, 41, 1943. (c) Shono, T.; Nishiguchi, I.; Omizu, H. *Chem. Lett.* 1976, 1233. (d) Shono, T.; Nishiguchi, I.; Ohmizu, H.; Mitani, M. *J. Am. Chem. Soc.* 1978, 100, 545. (e) Pradhan, S. K.; Kadam, S. R.; Kolhe, J. N.; Radhakrishnan, T. V.; Sohani, S. V.; Thaker, V. B. *J. Org. Chem.* 1981, 46, 2622. (f) Corey, E. J.; Pyne, S. G. *Tetrahedron Lett.* 1985, 24, 2821. (g) Fox, D. P.; Little, R. D.; Baizer, M. M. *J. Org. Chem.* 1985, 50, 2202. (h) Patten, G.; Robertson, G. M. *Tetrahedron* 1985, 41, 4001. (i) Belotti, D.; Cossy, J.; Pete, J. P.; Portella, C. *Tetrahedron Lett.* 1985, 26, 4591. (j) Ikeda, T.; Yue, S.; Hutchinson, C. R. *J. Org. Chem.* 1985, 50, 5193. (k) Kariv-Miller, E.; Mahachi, T. *J. J. Org. Chem.* 1986, 51, 1041. (l) Crandall, J. K.; Mualla, M. *Tetrahedron Lett.* 1986, 27, 2243. (m) Fukuzawa, S.; Iida, M.; Nakanishi, A.; Fujinami, T.; Sakai, S. *J. Chem. Soc., Chem. Commun.* 1987, 920. (n) Molander, G. A.; Kenny, C. *Tetrahedron Lett.* 1987, 28, 4367.

(2) Previous synthetic studies directed toward tunicaminylluracil: (a) Suami, T.; Sasai, H.; Matsuno, K. *Chem. Lett.* 1983, 819. (b) Corey, E. J.; Samuelsson, B.; Luzzio, F. A. *J. Am. Chem. Soc.* 1984, 106, 3682. (c) Corey, E. J.; Samuelsson, B. *J. Org. Chem.* 1984, 49, 4735. (d) Suami, T.; Sasai, H.; Matsuno, K.; Suzuki, N.; Fukuda, Y.; Sakana, O. *Tetrahedron Lett.* 1984, 25, 4533. (e) Danishesky, S.; Barbachyn, M. *J. Am. Chem. Soc.* 1985, 107, 7761. (f) Suami, T.; Sasai, H.; Matsuno, K.; Suzuki, N. *Carbohydr. Res.* 1985, 143, 85. (g) Danishesky, S. J.; DeNinno, S. L.; Chen, S.; Boisvert, L.; Barbachyn, M. *J. Am. Chem. Soc.* 1989, 111, 5810.

(3) The tunicamycins exhibit a spectrum of biological activity associated with their ability to inhibit glycoprotein synthesis in bacterial, fungal, mammalian, and viral systems. Isolation: (a) Takatsuki, A.; Arima, K.; Tamura, G. *J. Antibiot.* 1971, 24, 215. (b) Kenig, M.; Reading, C. *J. Antibiot.* 1979, 32, 549. (c) Hamill, R. L. U.S. Patents 4,237,225, 1980; 4,336,333, 1982. Structure elucidation of tunicaminylluracil: (d) Ito, T.; Kodama, Y.; Kawamura, K.; Suzuki, K.; Takatsuki, A.; Tamura, G. *Agric. Biol. Chem.* 1977, 41, 2303. Structure elucidation of tunicamycins: (e) Takatsuki, A.; Kawamura, K.; Okina, M.; Kodama, Y.; Ito, T.; Tamura, G. *Agric. Biol. Chem.* 1977, 41, 2307. (f) Takatsuki, A.; Kawamura, K.; Kodama, Y.; Ito, T.; Tamura, G. *Agric. Biol. Chem.* 1979, 43, 761. (g) Ito, T.; Takatsuki, A.; Kawamura, K.; Sato, K.; Tamura, G. *Agric. Biol. Chem.* 1980, 44, 695. Biological activity: (h) Takatsuki, A.; Tamura, G. *J. Antibiot.* 1971, 24, 224. (i) Takatsuki, A.; Tamura, G. *J. Antibiot.* 1971, 24, 785. (j) Takatsuki, A.; Kohno, K.; Tamura, G. *Agric. Biol. Chem.* 1975, 39, 2089. Reviews: (k) Elbein, A. *Trends Biochem. Sci.* 1981, 219. (l) *Tunicamycin*; Tamura, G., Ed.; Japan Scientific Press: Tokyo, Japan, 1982. (m) Eckardt, K. *J. Nat. Prod.* 1983, 46, 544.